

Modern Chemistry Chapter 7 Review

for an ionic compound represents one formula unit-- the simplest ratio of the compound's positives ions (cations) and its negative ions (anions).

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Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion. (d) the total positive charge of the ...

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